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Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713597286>

Chemical Route to the Capped RNAs

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Online publication date: 28 October 2004

To cite this Article Koukhareva, I. I. and Lebedev, A. V.(2004) 'Chemical Route to the Capped RNAs', Nucleosides, Nucleotides and Nucleic Acids, 23: 10, 1667 — 1680

To link to this Article: DOI: 10.1081/NCN-200031492

URL: <http://dx.doi.org/10.1081/NCN-200031492>

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Chemical Route to the Capped RNAs

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ABSTRACT

Eukaryotic and viral messenger RNAs contain a CAP structure that plays an important role in the initiation of translation and several other cellular processes that involve mRNAs. In this paper, we report a convenient chemical approach to the preparation of milligram quantities of short, capped RNA oligonucleotides, which overcomes some of the limitations of previous approaches. The method is based on the use of a reactive precursor, m⁷GppQ [P¹-7-methylguanosine-5'-O-yl, P²-O-8-(5-chloroquinolyl) pyrophosphate]. The precursor reacts smoothly with 5'-phosphorylated unprotected short RNA in the presence of CuCl₂ in organic media. The feasibility of this approach was demonstrated by the synthesis of the capped pentaribonucleotide m⁷GpppGpApCpU. The synthesized capped oligonucleotide was isolated and purified by reverse phase and ion exchange HPLC with a final yield of 37%. The structure of the m⁷GpppGpApCpU was confirmed by ³¹P NMR, mass-spectrometry and enzymatic hydrolysis.

Key Words: mRNA; 5'-capped mRNA; 5'-capped oligoribonucleotide; m⁷GpppG; Chemical synthesis of 5'-capped oligoribonucleotide.

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INTRODUCTION

Eukaryotic and viral messenger RNAs^[1–4] contain a CAP structure ($m^7GpppG...$) at the 5' end where 7-methylguanosine is attached to the mRNA through a 5'-5'-triphosphate bridge. In vivo the CAP structure is attached to an mRNA transcript when it reaches a chain length of 25–30 nucleotides^[5–7] by consecutive enzymatic reactions involving RNA 5'-triphosphatase, RNA guanylyltransferase and RNA guanine-7-methylase.^[2–4,8] Extensive studies conducted over the past 30 years have revealed that the CAP structure plays an important role in processes leading to the initiation of peptide synthesis on the ribosome (transcript processing, transport, translation, turnover).^[2–4,9]

Despite understanding the significance of the CAP structure in these cellular processes, reports of structural studies describing interactions between capped RNA oligonucleotides and different binding proteins, initiating factors, or ribosomal elements have been very limited.^[10–12] These types of studies, which use techniques such as crystallography, 2D NMR, calorimetric studies, and stop-flow applications, require milligram quantities of pure compounds. Current synthetic methods, whether enzymatic or chemical, cannot easily produce capped RNAs in sufficient quantities for the investigations which are needed to uncover the structural details of binding interactions.

Typically, capped short and long RNAs are prepared enzymatically in small quantities by transcription reactions with T3, T7 or SP6 polymerases using the chemically synthesized dinucleotide m^7GpppG as a primer.^[12] However, this approach results in the formation of two isomers of the capped RNAs with either natural ($m^7GpppG...$) or reverse ($Gpppm^7G...$) orientations that are present in ratios from 2:1 to 1:1.^[13]

Recently, new CAP analogs have been designed as a strategy to eliminate formation of isomers. Substitution of 7-methylguanosine in the m^7GpppG for 2'- or 3'-deoxy analogs (or 2'- or 3'-methoxy analogs) of 7-methylguanosine,^[14–16] when used as a primer for in vitro synthesis, resulted in production of capped RNAs only in the natural orientation. Another enzyme, gene 4 primase, uses m^7GpppA as a substrate in DNA-dependent synthesis of short RNA templates apparently exclusively in the natural direction.^[10] It was also demonstrated that synthetic RNA oligonucleotides containing 5'-diphosphate^[17] or 5'-triphosphate^[18] groups can be converted enzymatically to the desired capped oligomers. In general, the enzymatic approaches are suitable for small-scale synthesis (i.e., micrograms) of capped RNAs.

However, for the preparation of significant quantities of capped RNA oligonucleotides a versatile all-chemical route is needed. There have been several methods reported in the past 20 years for the synthesis of capped RNAs (Refs. [19–28] see Table 1). Generally, all approaches may be divided in two major schemes for the attachment of the cap structure to the RNA molecule. The most widely used approach relies on the activation of phosphate function of m^7G -containing precursor and the other method uses activation of the terminal 5'-phosphate function of the RNA oligonucleotide.

In most cases activation of the m^7G -precursor was achieved through formation of an imidazolidine of the 5'-phosphomonoester or pyrophosphate 5'-monoester derivative of m^7G (Im- pm^7G or Im- ppm^7G , respectively). The reaction is facilitated by catalysis with metal ions such as Mn^{2+} , Mg^{2+} , Ca^{2+} (for aqueous solutions) or Ag^+ , Cu^{2+} (for non-aqueous media). Typically, in aqueous media the reaction requires 6–10 days for completion, and in organic solution the reaction proceeds 5–10 times faster.

Table 1. Chemically synthesized capped RNA oligonucleotides.

Capped RNA oligonucleotide sequence	Activated molecule	m ⁷ G-precursor ^a	Starting RNA ^b	Catalyst	Conversion % ^c	Isolated yield % ^d (scale)	Ref.
<i>New Procedure</i>							
m ⁷ GpppGACU	m ⁷ G precursor	m ⁷ GppQ	U/P	CuCl ₂	80%	37% (1.5 μmole)	This work
<i>Published</i>							
m ⁷ GpppAUG	m ⁷ G precursor	ArS-ppm ⁷ G (S, B)	T, B	AgNO ₃ /imidazole	40–50%	8% (0.5 μmole)	[19,20]
m ⁷ GpppAUGACC	m ⁷ G precursor	ArS-ppm ⁷ G (S, B)	T, B	AgNO ₃ /imidazole	40–50%	5% (0.25 μmole)	[19,20]
m ⁷ GpppGUAUUA	5'-Im-pRNA	ppm ⁷ G (S, B)	T	Not used	> 50%	16% (0.08 μmole)	[21–23]
m ⁷ GpppGUAUUAUA	5'-Im-pRNA	ppm ⁷ G (S, B)	T	Not used	NR	5% (0.2 μmole)	[21,24]
m ⁷ GpppGUA	m ⁷ G precursor	ArS-ppm ⁷ G (S, B)	T	AgNO ₃ /imidazole	NR	11% (0.41 μmole)	[23]
m ⁷ GpppGUAUUA	m ⁷ G precursor	ArS-ppm ⁷ G (S, B)	T	AgNO ₃ /imidazole	NR	2% (0.06 μmole)	[23]
m ⁷ GpppAAAAA	m ⁷ G precursor	Im-ppm ⁷ G	U/P	MnCl ₂	49%	NR (NR)	[25]
m ⁷ GpppACACUUGCUUU	m ⁷ G precursor	Im-ppm ⁷ G	U/P	MnCl ₂	35%	NR (NR)	[25]
m ₃ ^{2,2,7} GpppAmUmA	m ₃ ^{2,2,7} G precursor	Im-pm ₃ ^{2,2,7} G	M	Not used	NR	40% (8.8 μmole)	[26]
m ₃ ^{2,2,7} GpppAmUmA	m ₃ ^{2,2,7} G precursor	Im-pm ₃ ^{2,2,7} G (S)	M, T	Not used	30–40%	20% (NR)	[27]
m ₃ ^{2,2,7} GpppAmUmAm	5'-Im-pRNA	ppm ⁷ G	M	ZnCl ₂	NR	NR (NR)	[28]

^aB–N2-base protected 7-methylguanosine, S–sugar protected.^bT–2'-THP protected nucleoside(s), B–base protected nucleoside(s), M–2'-OMe nucleosides, U/P–unprotected oligomer.^cPercentage of starting RNA oligomer converted to capped RNA oligomer (estimated from HPLC profile).^dIsolated yield of capped RNA oligomer (NR–not reported).

Activation of 5'-phosphate function of RNA was carried out by converting to a 5'-phosphoroimidazole derivative using coupling reagents (such as N,N-carboxyldiimidazole or triphenyl phosphine/ α,α -dipyridyl-disulfide). For reactions in organic media, RNA oligonucleotides containing acid labile 2'-protecting groups compatible with CAP structure and RNA backbone (or 2'-O-Methyl RNA derivatives) were used. The activated RNA oligomer was then reacted with pp-m⁷G and once the reaction was completed the 2'-protecting groups were removed by treatment with 0.01M HCl (pH 2). Those syntheses required an additional deprotection step and an extensive chromatography purification was needed to isolate pure products. The overall yield of this process was quite low.

Despite numerous methods developed for preparing capped RNAs, the substantial amounts needed for detailed investigations are still not readily available. On the other hand, a large scale chemical synthesis of m⁷GpppG was described ten years ago by Hata and co-workers.^[23] The key intermediate P¹-guanosine-5'-O-yl, P²-O-8-(5-chloroquinolyl) pyrophosphate (GppQ) was reacted without isolation in the presence of CuCl₂ with pm⁷G to give the desired product, m⁷GpppG. It was later shown that similar reaction of isolated and purified P¹-7-methylguanosine-5'-O-yl, P²-O-8-(5-chloroquinolyl) pyrophosphate (m⁷GppQ) with pA produced m⁷GpppA with a moderate yield.^[10] It was logical to approach the synthesis of the capped RNA oligonucleotides using a reaction of active precursor (m⁷GppQ) with a purified 5'-phosphorylated unprotected RNA oligonucleotide. Here, a convenient all-chemical approach to the synthesis of short, capped RNA oligonucleotides in quantities sufficient for many structural studies is described.

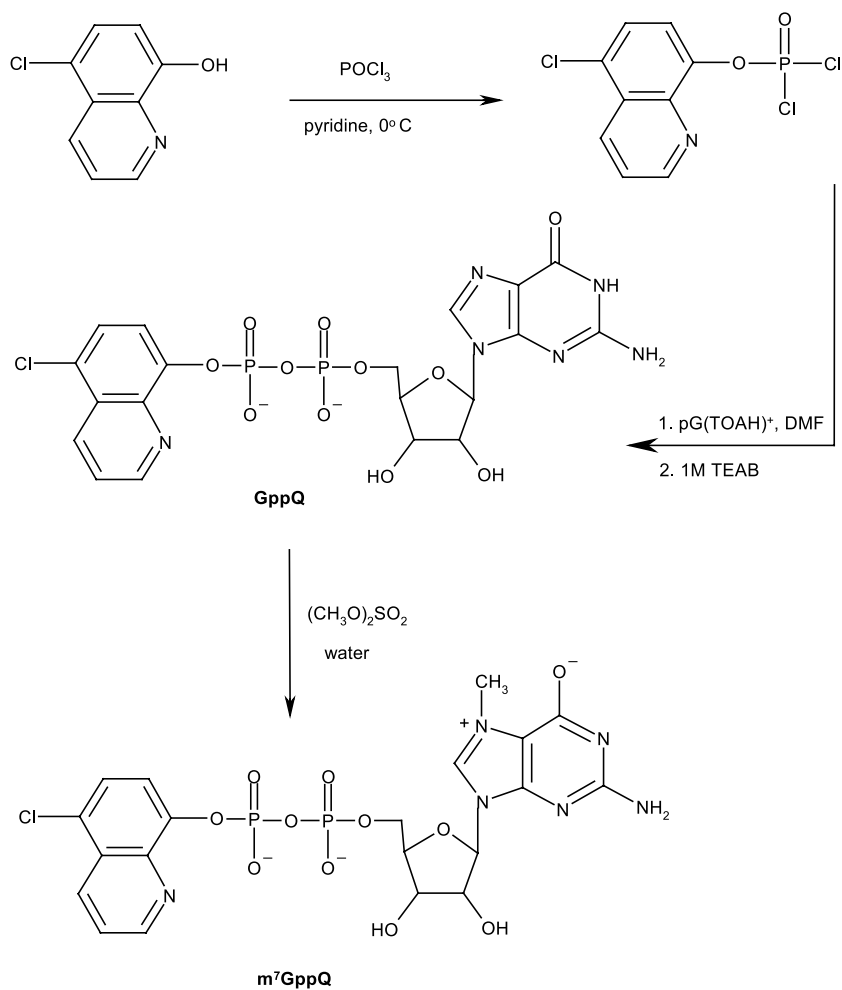
RESULTS

First, we tried a different procedure for preparation of m⁷GppQ. The treatment of 5-chloro-8-hydroxyquinoline in pyridine with an excess of POCl₃^[29–31] leads to formation of the reactive intermediate, presumably 5-chloroquinolyl-8-dichlorophosphate (Scheme 1). The latter compound without isolation was reacted with the mono-tri-n-octylammonium salt of guanosine 5'-monophosphate to give P¹-guanosine-5'-O-yl, P²-O-8-(5-chloroquinolyl) pyrophosphate (GppQ) in 30% yield after isolation by DEAE anion-exchange chromatography.

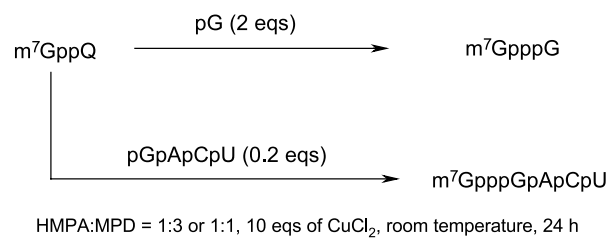
The GppQ was further treated with dimethyl sulfate in water for several hours to produce m⁷GppQ. It was isolated after purification with reverse phase chromatography with 67% yield as a triethylammonium salt. The overall yield of 20% was comparable to 35% yield obtained in Ref. [10] where a different synthetic route for preparation of m⁷GppQ was used.

To test the efficiency of the m⁷GppQ prepared in this work we synthesized the simplest capped RNA derivative, m⁷GpppG (Scheme 2), by reacting m⁷GppQ with pG in the presence of CuCl₂ in the mixture of hexamethylphosphoramide (HMPA) and 1-methyl-2-pyrrolidinone (MPD) (1:3, v/v). HPLC analysis (not presented) of the reaction showed the main product formed was identical to an authentic sample of m⁷GpppG.

However, the feasibility of the above approach for the synthesis of capped RNA oligonucleotides by a reaction of m⁷GppQ with a purified 5'-phosphorylated



Scheme 1. Chemical synthesis of m⁷GppQ.



Scheme 2. Chemical synthesis of m⁷GpppG and m⁷GpppGpApCpU.

unprotected RNA oligonucleotide had to be proven. One significant issue was the unknown solubility of RNA oligomers in HMPA-MPD organic media in the presence of CuCl_2 . Also, the ability of phosphodiester bonds of unprotected RNA oligomers to withstand exposure to CuCl_2 in HMPA-MPD organic media for prolonged periods of time (hours or days) was not known. This was a considerable concern given the well-known ability of metals to catalyze RNA hydrolysis.^[32]

These questions were answered in the following experiment. A 14-mer RNA oligonucleotide was chosen as a model for stability studies to represent intermediate length oligonucleotides. The 14-mer was first converted to the cetyl-triethylammonium (CTMAH⁺) salt using a published procedure^[33] in order to ensure solubility at a concentration of at least 1 mM at room temperature. The 24 h treatment of the 14-mer with 200 molar excess of CuCl_2 in HMPA-MPD (1:1) revealed some degradation of phosphodiester linkages (Fig. 1). However, at this point we considered the level of

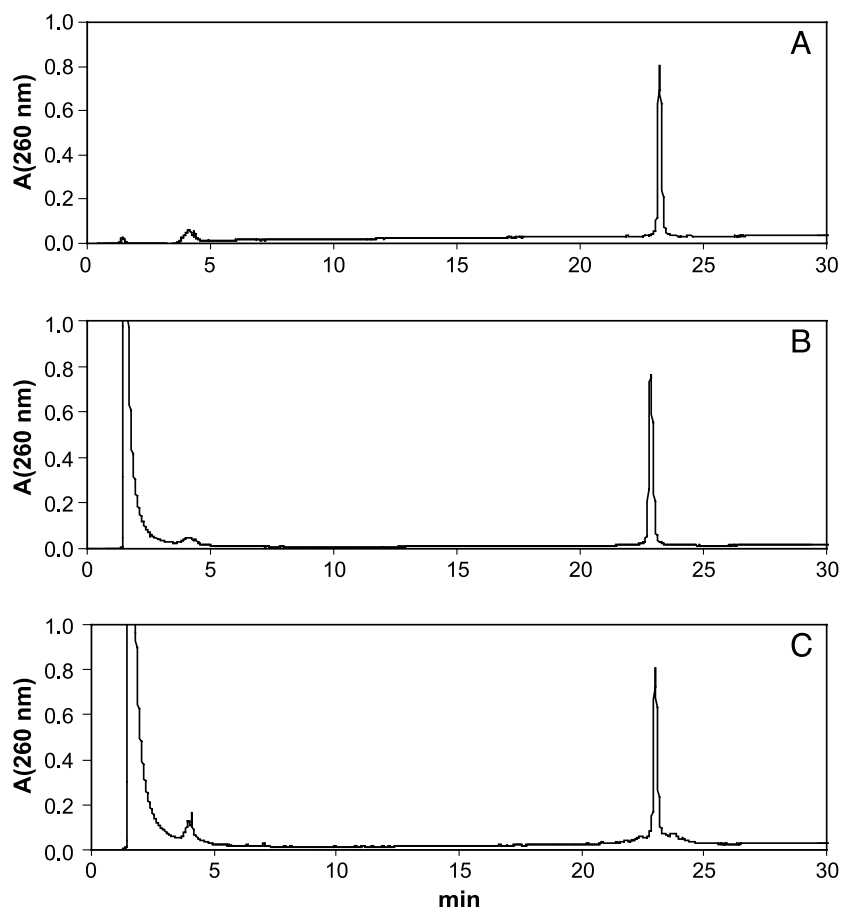


Figure 1. Anion-exchange HPLC analyses of 14-mer RNA (1 mM) dissolved in HMPA:MPD (1:1) mixture in the absence of CuCl_2 (A) and after incubation in the presence of 200 mM CuCl_2 for 2 hours (B) and 24 hours (C).

RNA degradation to be acceptable, especially since a much lower concentration (and excess) of CuCl_2 would be used in the actual capping reaction (see below).

Finally, a short 4-mer RNA containing all four common RNA nucleosides (A, G, C and U) was chosen to evaluate the capping reaction with $m^7\text{GppQ}$ in the presence of CuCl_2 . The choice of the length and structure of the model 4-mer RNA and the scale of the synthesis was dictated by the necessity to simplify the isolation of the final product and confirm its structure. $p\text{GpApCpU}$ was synthesized on the solid support using standard phosphoramidite chemistry. The oligonucleotide was deprotected and purified by reverse phase HPLC and DEAE Sephadex A25 anion-exchange chromatography and isolated as a triethylammonium salt.

To our surprise, the triethylammonium salt of $p\text{GpApCpU}$ was soluble in the HMPA-MPD (1:1) mixture in the presence of CuCl_2 . Therefore, conversion of the 4-mer RNA to the CTMAH^+ salt was not necessary. The results of the reaction of the $m^7\text{GppQ}$ and $p\text{GpApCpU}$ (see Scheme 2) are illustrated in the HPLC chromatograms (Fig. 2). Analysis of the reaction mixture after 2 hours revealed a formation of the new compound, presumably $m^7\text{GpppGpApCpU}$, which had an HPLC retention time approximately 1 minute longer than that of the starting material, $p\text{GpApCpU}$. The $m^7\text{GpppGpApCpU}$ became the major oligonucleotide component of the mixture after the reaction proceeded for 23 hours. At this point the mixture was diluted with water, and $m^7\text{GpppGpApCpU}$ was isolated, and purified by reverse phase and ion-exchange

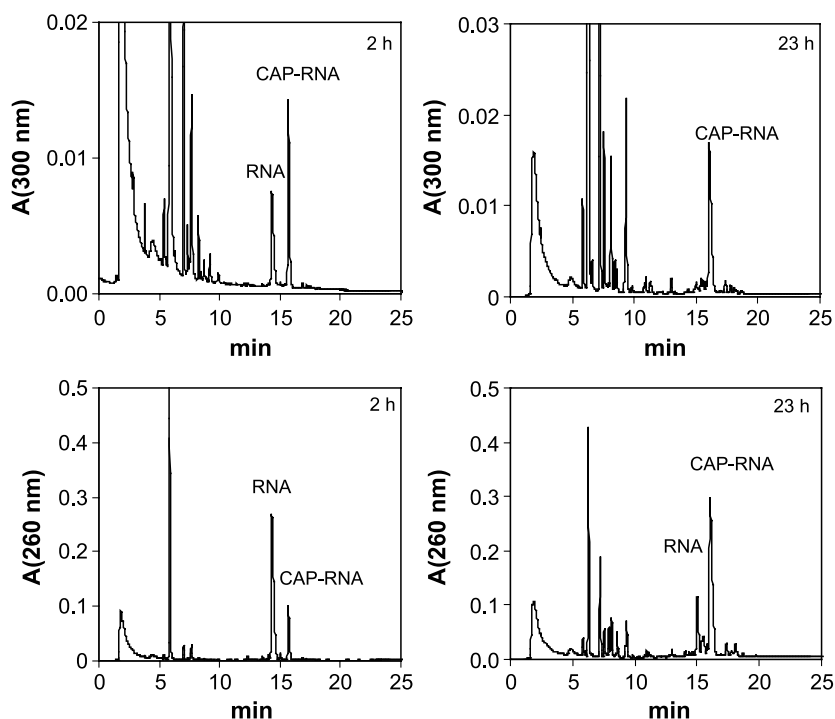


Figure 2. Synthesis of $m^7\text{GpppGpApCpU}$ (CAP-RNA): anion-exchange HPLC analyses of the reaction mixture at 2 and 23 hours (detection at 260 or 300 nm).

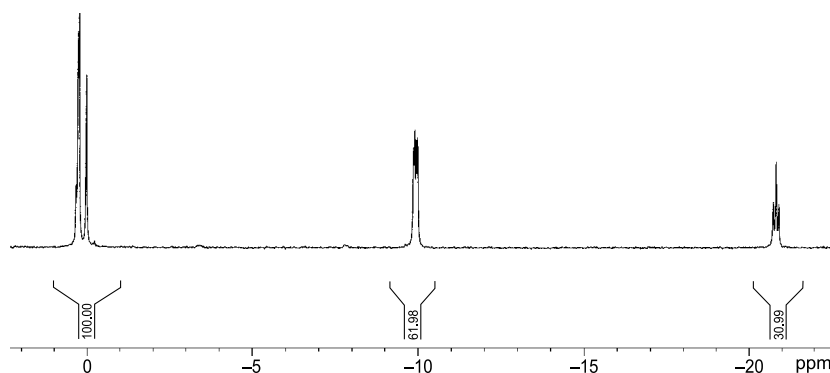


Figure 3. $^{31}\text{P}\{-^1\text{H}\}$ NMR spectrum of $\text{m}^7\text{GpppGpApCpU}$ in D_2O .

HPLC. Based on the quantity of the starting 4-mer RNA oligonucleotide, a final yield of 37% ($\text{m}^7\text{GpppGpApCpU}$) was obtained.

The structure of isolated $\text{m}^7\text{GpppGpApCpU}$ was confirmed using four different methods of analysis. First, while the starting pGpApCpU does not have significant absorbance at > 300 nm, absorbance of the $\text{m}^7\text{GpppGpApCpU}$ at 300 nm is relatively high, indicating the presence of m^7G (at pH 10, the $A_{300\text{ nm}}/A_{260\text{ nm}}$ ratios are 0.029 and 0.136 for pGpApCpU and $\text{m}^7\text{GpppGpApCpU}$ respectively; see Fig. 2). Second, the MALDI mass-spectrum of the purified product confirmed the molecular mass. Third, the ^{31}P NMR spectrum (Fig. 3) contains all expected phosphorus resonances. Finally, the enzymatic hydrolysis of $\text{m}^7\text{GpppGpApCpU}$ gave the expected ratio of all nucleotides products ($\text{pC:pU:pA:m}^7\text{GpppG}$) in the correct proportion (not shown). The successful enzymatic hydrolysis also serves as evidence that no noticeable isomerization of the phosphodiester internucleotide bonds ($3'-5'$ to $2'-5'$) took place during the synthesis and isolation of $\text{m}^7\text{GpppGpApCpU}$.

DISCUSSION

The first chemical syntheses of capped RNA oligomers were performed by Sekine and co-workers by the reaction of $\text{P}^1\text{-5'-O-(7-methylguanosine)}$, $\text{P}^2\text{-S-(4-methoxyphenyl)}$ pyrophosphate with 5'-phosphorylated/2'-THP protected RNA oligonucleotides in the presence of silver nitrate and imidazole.^[19,20] The reaction proceeded with up to 40–50% conversion efficiency. Following acidic treatment and purification, $\text{m}^7\text{GpppApUpG}$ and $\text{m}^7\text{GpppApUpGpApCpC}$ were isolated with 8% and 5% yields, respectively. Later some improvement in the yield of capping reaction was achieved when 5'-phosphorylated RNA oligonucleotide (2'-THP protected) was converted to respective 5'-phosphoroimidazole and reacted with ppm^7G (N2 and 2', 3'-protected).^[21,22]

Sawai and co-workers achieved a similar efficiency of capping of RNA oligomers.^[25] They prepared $\text{Im-ppm}^7\text{G}$ and used it to react with deprotected and purified 5'-phosphorylated 6-mer and 11-mer oligoribonucleotides in aqueous solution in the presence of Mn^{2+} ions (reaction in the presence of Mg^{2+} or Ca^{2+} was less efficient). The process required 10-fold excess of the $\text{Im-ppm}^7\text{G}$ over oligonucleotide

and continued for 6–10 days at 25–30°C. Conversions of 49% (6-mer) and 35% (11-mer) to the capped oligonucleotides were achieved. The yield was calculated by integration of the respective peaks of starting oligonucleotide and capped oligonucleotide from analytical HPLC. It was demonstrated that the hydrolysis of Im-ppm⁷G in 250 mM tris-acetate buffer (pH 7.5) seems to be slow (half-life time was estimated to be 142 hours at 25°C). However, it remains unclear if the unavoidable hydrolysis of Im-ppm⁷G is accelerated in the presence of Mn²⁺.

Recently, Darzynkiewicz and co-workers reported the efficient Zn²⁺ catalysis of the reaction of Im-pm⁷G or Im-ppm⁷G with ppG or pG, respectively, in organic media (DMF).^[14] They used this approach for preparation of m⁷GpppAmUmAm although no details of the reaction conditions, time of the reaction and no isolated yield were reported.^[28]

In this publication, we report a convenient and reliable method for the synthesis of milligram quantities of short, capped RNA oligonucleotides (the capping experiments on pGpApCpU were successfully repeated three times). The method relies on the use of stable compounds: fully deprotected 5'-phosphorylated RNA oligonucleotide and m⁷GppQ. We suggest a simple and easy way to prepare the key m⁷G-precursor, m⁷GppQ. It was demonstrated that trialkylammonium and tetraalkylammonium salts of 5'-phosphorylated and fully unprotected short oligonucleotides are soluble enough in HMPA-MPD (1:1) media to support an efficient capping reaction. The level of RNA oligonucleotide degradation during capping reaction is noticeable but considered acceptable for short RNA oligonucleotides and no significant isomerization of internucleotide phosphodiester linkages occurred. Also, the opportunity to use the pre-synthesized, fully deprotected and purified RNA oligomers significantly facilitates the purification and isolation of the final capped RNA oligonucleotide. Activation of m⁷GppQ with CuCl₂ proceeds without noticeable complications and the RNA capping reaction could be completed in 23 hours. The isolated yield of capped RNA after purification by reverse phase and ion exchange chromatography was nearly 37% (based on starting RNA oligonucleotide).

Work is currently in progress on the optimization of conditions of capping reaction for RNA oligonucleotides of the intermediate length (10–20 mers). Also, an easier and more convenient method of preparing m⁷GppQ by reaction of 5-chloroquinolyl-8-dichlorophosphate with pm⁷G similar to preparation of GppQ (see above) is in development.

MATERIALS AND METHODS

Guanosine 5'-monophosphate (free acid) was purchased from ChemImpex, copper (II) chloride (anhydrous) from Acros Organics. Polymer support, fully protected nucleoside 3'-phosphoramidite monomers and 5'-phosphorylating reagent were purchased from Glen Research. Other chemicals and solvents were purchased from Sigma-Aldrich or Fisher Scientific and used without further purification.

NMR spectra (¹H, ¹³C and ³¹P) were recorded on Bruker AMX 500 instrument (NuMega, San Diego, CA) in D₂O at 25°C. Mass spectra were recorded on Applied Biosystems's instrument Voyager DE (HT Laboratories, San Diego, CA). HPLC (preparative and analytical) were performed on Beckman Gold HPLC System equipped with diode array UV detector unit (Beckman Instruments). For anion exchange HPLC a

gradient of 1M LiCl in 25 mM Tris-base (pH 10.0) was used; for reverse phase HPLC a gradient of acetonitrile in 100 mM triethylammonium acetate (pH 7.2) was used.

Oligoribonucleotides were synthesized (GCC ACG CCC CGA UU in 0.2 μ mole scale and pGACU in 15 μ mole scale) on polymer support on Expedite 8909 (PerSeptive Biosystems) using standard manufacturer suggested protocols for RNA synthesis. The oligoribonucleotides were deprotected and removed from polymer support using standard concentrated ammonia/tetrabutylammonium fluoride deprotection procedure.^[34] The crude 14-mer was purified with preparative PAGE (then eluted and ethanol precipitated as Na⁺ salt). The 4-mer was isolated by preparative reverse phase HPLC in gradient of acetonitrile (0–50%) in 100 mM triethylammonium acetate and then purified by ion exchange DEAE Sephadex A25 chromatography in gradient of triethylammonium bicarbonate, pH 7.5 (0.02 M to 1.0 M). Oligonucleotides were at least 95% pure by analytical anion exchange and reverse phase HPLC.

Synthesis of m⁷GppQ

Preparation of Mono-tri-n-octylammonium Salt of Guanosine 5'-Monophosphate

To a stirring suspension of guanosine 5'-monophosphate (908 mg; free acid; 2.5 mmol) in 5 mL of methanol added 1.08 mL of tri-n-octylamine (876 mg; 2.5 mmol). After 10 min of stirring at room temperature the clear solution was evaporated to white foam. The foam was dissolved in 6 mL of dry DMF at 50°C and evaporated again to white foam and put on high vacuum for 24 h. The foam was dissolved in 5 mL of dry DMF (to be used in next step).

Preparation of GppQ

A solution of 5-chloro-8-hydroxyquinoline (385 mg; 2.14 mmol) in 2 mL of anhydrous pyridine was added dropwise to a solution of phosphorus oxychloride (600 μ L; 6.44 mmol) in 3 mL of anhydrous pyridine with stirring and cooling in an ice bath. The bath was removed and the mixture was stirred for 5 h at room temperature. The pyridinium chloride was filtered off and the filtrate was evaporated to a yellow oily residue. The obtained 5-chloroquinolyl-8-dichlorophosphate was dissolved in 5 mL of dry DMF. To this mixture was added the solution of guanosine 5'-monophosphate (TOAH⁺ salt, 5mL in DMF) and 510 μ L of tri-n-butylamine with stirring. After 10 min at room temperature the reaction was quenched with 5 mL of 1 M TEAB (pH 7.5). The mixture was evaporated down to an oily residue and then shaken with dichloromethane:water = 1:1 (v/v, 200 mL). The organic layer was washed twice with 100 mL of water; the aqueous layers were combined and loaded on column (2.6 \times 90 cm) with Sephadex-A25 DEAE (bicarbonate form, equilibrated with water). The column was washed with 0.5 L of water and the product was eluted with a linear gradient of 0.0–1.0 M TEAB (2L). Fractions containing the product with purity higher then 90% (determined by anion exchange HPLC) were combined, evaporated to a white foam and put on high vacuum for 24 hrs. The isolated yield was 520 mg (30%). ³¹P-{¹H} NMR (D₂O), δ : -10.79 (d) and -16.62 (d); J_{POP} 20.6 Hz. ¹³C-{¹H} NMR (D₂O), δ : 157.04, 152.28, 149.89, 149.07, 145.09 (d), 138.76 (d), 136.13, 132.65, 125.57, 125.39, 124.06,

121.60, 117.00, 114.75, 86.26, 82.36 (d), 72.47, 69.11, 64.25 (d), 45.65 (CH₂; TEAH⁺), 7.22 (CH₃; TEAH⁺). Electrospray mass spectrum: 603 and 605 (3:1; M – H⁺). UV (water): λ_{\max} 239 nm; λ_{\min} 222 nm; λ_{sh} 247, 275, 310 nm.

Methylation of GppQ

Dimethyl sulfate (8.6 μL , 90 μmol) was added to a solution of GppQ (73 mg, 103 μmol , TEAH⁺ salt) in 300 μL of water and the reaction mixture was stirred at room temperature for 4 hrs. The m⁷GppQ was isolated by preparative reverse phase HPLC on 25 \times 100 mm C18 cartridge (Waters) in gradient of acetonitrile in 100 mM TEAA, pH 7.2 (0–50% over 50 min, 5 mL/min). Fractions containing m⁷GppQ were evaporated to remove acetonitrile and the product was re-purified and de-salted on the same cartridge with a linear gradient of acetonitrile in water (0–50% over 20 min). Fractions containing the m⁷GppQ were combined, evaporated to white foam and dried on high vacuum for 24 hrs. The isolated yield was 50 mg (67%). ³¹P-{¹H} NMR (D₂O), δ : –10.65 (d) and –16.65 (d); J_{POP} 20.1 Hz. ¹H NMR (D₂O), δ : 8.819 (1H, dd); 8.463 (1H, dd); 7.573 (1H, dd); 7.511 (2H, s); 5.747 (1H, d); 4.40–4.47 (3H, m); 4.342 (1H, m); 4.221 (1H, m); 3.633 (3H, s); 3.175 (6H, q, CH₂; TEAH⁺); 1.256 (9H, t, CH₃; TEAH⁺). Electrospray Mass spectrum: 617 and 619 (3:1; M – H⁺); 619 and 621 (3:1; M + H⁺). UV (water): λ_{\max} 239, 257 and 284 nm; λ_{\min} 224, 253, and 277 nm; λ_{sh} 317 nm.

Reaction of m⁷GppQ with pG (Analytical Experiment)

Approximately 1 mg of pG (TOAH⁺ salt; \sim 2 μmol) was mixed with 2 equivalents of m⁷GppQ (TEAH⁺ salt) in 100 μL of HMPA:MPD = 1:3 (v/v) in 1.5 mL conical tube. Solid CuCl₂ (5.4 mg; 40 μmol) was added to the reaction mixture and the tube was vortexed at room temperature. The progress of the reaction (up to 40 h) was monitored by anion exchange HPLC (identification of m⁷GpppG was performed by the comparison with authentic sample: commercial product, TriLink BioTechnologies, Inc.). Note: m⁷GpppG is relatively stable (for at least 20 min) during anion exchange HPLC analysis in 25 mM Tris-base (pH 10.0).

Synthesis of m⁷GpppGpApCpU

To a solution of pGpApCpU (163 O.D. at 260 nm, 4.1 μmol , TEAH⁺ salt) in 650 μL of HMPA:MPD = 1:1 (v/v) was added solid m⁷GppQ, (14.5 mg, 20 μmol , TEAH⁺ salt) followed by solid CuCl₂ (6.0 mg, 44.6 μmol). The mixture was kept at room temperature on vortex and the progress of the reaction was monitored by anion exchange HPLC. After 24 hrs the mixture was poured into 100 mL of water and the resulting solution was purified immediately by reverse-phase HPLC on 25 \times 100 mm C18 cartridge (Waters) in gradient of acetonitrile in 100 mM TEAA, pH 7.2 (0–30% over 50 min, 5 mL/min). Fractions containing m⁷GpppGpApCpU were combined, evaporated to remove acetonitrile and the product was de-salted on the same cartridge in gradient of acetonitrile in water (0–30% over 50 min, 5 mL/min). m⁷GpppGpApCpU was further HPLC purified using anion-exchange Resource Q 8 mL column (Pharmacia) in gradient of 1 M LiCl in water (0–25% over 40 min, 5 mL/min) and de-salted as above to give 69 O.D. (at 260 nm; \sim 1.5 μmol). ³¹P-{¹H} NMR (D₂O), δ : 0.283 (1P, s), 0.252 (1P, s), 0.028 (1P, s); –9.895 (1P, d, J_{pop}

17.2 Hz); -9.941 (1P, d, J_{pop} 17.2 Hz); -20.822 (1P, t, J_{pop} 17.2 Hz). MALDI Mass-spectrum: 1744 ($M + H^+$).

Enzymatic Hydrolysis of $m^7\text{GpppGpApCpU}$ with Mung Bean Nuclease

To a solution of $m^7\text{GpppGpApCpU}$ in water (1.0 O.D. at 260 nm; 10 μL) was added 1.4 μL of MBN 10 \times buffer (BioLabs, New England) and 3.0 μL of MBN enzyme (10 U/ μL ; BioLabs, New England). The mixture was incubated at 37°C and 2 μL aliquots were removed and analyzed by reverse phase HPLC. Hydrolysis of $m^7\text{GpppGpApCpU}$ was completed in several hours. The peaks corresponding to nucleotides pA, pC, pU and $m^7\text{GpppG}$ were identified by UV spectra and by co-injection with authentic samples.

Stability Study of 14-mer RNA Oligonucleotide in the Presence of CuCl_2

The 14-mer oligonucleotide GCC ACG CCC CGA UU (Na^+ salt) was converted to cetyltrimethylammonium (CTMAH^+) salt according to a published procedure.^[28] The residue (7.3 O.D. at 260 nm, 58 nmole) was dried down on high vacuum for 12 hours and dissolved in 58 μL of the mixture HMPA:MPD = 1:1 (v/v) in 1.5 mL conical tube. Solid CuCl_2 (0.5 mg; 4 μmole) was added to a 20 μL aliquot of that solution and the mixture was vortexed at room temperature. The integrity of 14-mer was monitored by anion exchange HPLC (Fig. 1).

ACKNOWLEDGMENTS

This work is supported by the TriLink BioTechnologies, Inc. research program. We thank Dr. Richard I. Hogrefe and Dr. Morteza Vaghefi for interest, support and helpful discussions, Paul Imperial and Katelyn Hanesana for providing the synthesis of oligoribonucleotides, and Rose Murphy and Athena Krutka for manuscript editing.

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Received January 23, 2004

Accepted June 7, 2004